Mapping pollution exposure and chemistry during an extreme air quality event (the 2018 Kīlauea eruption) using a low-cost sensor network

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Extreme air quality episodes represent a major threat to human health worldwide but are highly dynamic and exceedingly challenging to monitor. The 2018 Kīlauea Lower East Rift Zone eruption (May to August 2018) blanketed much of Hawai‘i Island in “vog” (volcanic smog), a mixture of primary volcanic sulfur dioxide (SO2) gas and secondary particulate matter (PM). This episode was captured by several monitoring platforms, including a low-cost sensor (LCS) network consisting of 30 nodes designed and deployed specifically to monitor PM and SO2 during the event. Downwind of the eruption, network stations measured peak hourly PM2.5 and SO2 concentrations that exceeded 75 μg m−3 and 1,200 parts per billion (ppb), respectively. The LCS network’s high spatial density enabled highly granular estimates of human exposure to both pollutants during the eruption, which was not possible using preexisting air quality measurements. Because of overlaps in population distribution and plume dynamics, a much larger proportion of the island’s population was exposed to elevated levels of fine PM than to SO2. Additionally, the spatially distributed network was able to resolve the volcanic plume’s chemical evolution downwind of the eruption. Measurements find a mean SO2 conversion time of ∼36 h, demonstrating the ability of distributed LCS networks to observe reaction kinetics and quantify chemical transformations of air pollutants in a real-world setting. This work also highlights the utility of LCS networks for emergency response during extreme episodes to complement existing air quality monitoring approaches.

Significance

Poor air quality is a global public health issue, contributing to millions of premature deaths per year worldwide. Low-cost air quality sensors are a promising tool to improve monitoring capabilities. In this study, we built and deployed a low-cost sensor network for emergency response during an extreme air quality event, the 2018 Kīlauea Lower East Rift Zone eruption. This network was used to estimate fine-scale population exposure to multiple pollutants, to measure the chemical transformation of volcanic emissions, and to provide real-time observations as part of emergency management efforts.


The authors declare no competing interest.

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were destroyed, and thousands of residents were displaced. The elevated SO₂ emissions led to exceedingly poor AQ not only in the immediate vicinity of the eruption but also across the wider region. The order-of-magnitude step change in SO₂ emissions and resulting secondary PM was clearly visible from space (Fig. 1 A and B) and was also measured by the ground-based regulatory network (Fig. 1 C and D).

The measurements in Fig. 1 established the LERZ eruption as an extreme AQ event, and recent studies have used satellite and in situ measurements to explore air quality implications and plume dynamics during the eruption. For example, analysis of regulatory network data found that 24-h average PM₂.₅ concentrations exceeded US Environmental Protection Agency (EPA) AQ thresholds eight times during the eruption in certain locations, compared to zero times during the previous 8 y (17). However, such measurements are typically designed to monitor regional-scale AQ and therefore provide limited details about the fine-scale spatiotemporal distribution of air pollutants. Satellite measurements are limited both spatially and temporally (because of overpass intervals of 1 to 3 d, pixel sizes of ~tens of km, cloud cover, and limited vertical resolution). Ground-based regulatory measurements provide improved temporal resolution and networks are strategically placed to monitor ambient AQ in populated regions but are generally not designed to monitor fine-scale exposure from dynamic plumes during extreme events. On the Island of Hawai‘i, the average resident lived ~17 km from the nearest regulatory AQ station (closer than the United States average of 22 km), and while this network provides continuous, high-quality measurements at key locations, this is too sparse for high-resolution estimates of residents’ pollutant exposure given the high temporal and spatial variability of the volcanic plume.

**Results and Discussion**

In order to complement existing measurements and provide improved estimates of the pollutants’ spatial variability, human exposures, and rate of interconversion, we built and deployed a network of low-cost sensor (LCS) nodes to measure SO₂ and PM throughout the region. The relative affordability and small size of LCS enables many network nodes to be deployed within a small area, thereby providing distributed measurements with a much higher spatial resolution than is possible with traditional AQ networks (18–20). LCS networks have recently been deployed in different locations worldwide (20–23); while LCSs are generally less accurate and precise than regulatory AQ instruments, recent work has demonstrated that LCS calibration via colocation with regulatory-grade monitors can enable robust measurements (24–26). Our deployment during the eruption occurred in two phases. First, an initial small-scale deployment in the LERZ, enabling civil authorities and local residents to monitor SO₂ levels and make emergency management decisions, was carried out beginning May 14. These sensors had already been built and used in Hawai‘i (24) and were among the first SO₂ measurements in the area. Next, a total of 30 new sensor units were built (measuring SO₂ and PM), calibrated using the regulatory stations’ measurements, and deployed throughout the Island beginning May 23. Details of the sensor unit design and calibration are provided in the Materials and Methods and SI Appendix. After this deployment, the network had one node per ~5,800 people and the average resident lived 4.6 km from an AQ measurement (SI Appendix).

By comparison, the US- and globally averaged distances to a regulatory-grade AQ measurement are 22 km and 220 km, respectively (2).
Average SO$_2$ and PM$_{2.5}$ levels measured by a subset of the network during July 15 to August 1 are shown in Fig. 2. This subset (20 PM$_{2.5}$ and 17 SO$_2$ sensors) includes all sensors that operated nearly continuously throughout this measurement period (the full timeseries is shown in the SI Appendix). Both the regulatory and LCS networks measured similar overall trends in volcanic pollution downwind of the volcano; the highest SO$_2$ concentrations were observed just downwind of the eruption, whereas the highest PM$_{2.5}$ levels were further downwind, along the western (Kona) coast. LCS network stations measured peak hourly SO$_2$ concentrations $>$1,200 parts per billion (ppb) and peak hourly PM$_{2.5}$ concentrations $>$75 $\mu$g $\cdot$ m$^{-3}$ (SI Appendix).

The LCS network also revealed pollution gradients in populated areas that were imperceptible to the regulatory network, with substantial variability on finer (5 to 20 km) scales. For example, along the Kona coast, average PM$_{2.5}$ loadings at nearby sites varied by more than a factor of two (from 9 to 25 $\mu$g $\cdot$ m$^{-3}$) during the measurement period, likely because of differences in elevation, topography, and local winds. Sensors were also placed in areas that previously had been unaffected by volcanic pollution downwind of the volcano; the highest SO$_2$ concentrations were observed just downwind of the eruption, whereas the highest PM$_{2.5}$ levels were further downwind, along the western (Kona) coast. LCS network stations measured peak hourly SO$_2$ concentrations $>$1,200 parts per billion (ppb) and peak hourly PM$_{2.5}$ concentrations $>$75 $\mu$g $\cdot$ m$^{-3}$ (SI Appendix).

The high spatial resolution of the LCS network also enables fine-grained estimates of residents’ exposure to pollutants in terms of both average exposures (Fig. 3A and B) and hourly exposure distributions (Fig. 3C and D). Fine-grained exposure estimates have long been viewed as a primary advantage of distributed sensor networks (e.g., ref. 18), but there exist few, if any, examples of such networks used to quantitatively estimate population-wide exposure distributions. With this sensor-based analysis (covering the >70,000 people within 5 km of a sensor node), we are able to resolve the fine structure of pollutant exposure during this extreme AQ event, analogous to previous global population-scaled annual exposure estimates based on satellite data and models (27). The choice of a 5-km buffer is somewhat arbitrary but is intended as a compromise between population coverage (SI Appendix) and spatial variability of point measurements as the plume chemically transforms and dynamically adjusts to underlying surface conditions. During the eruption, the highest SO$_2$ levels were experienced primarily by those just downwind of the vents (Fig. 3A), with 5.3% of the sampled population exposed to elevated (>35 ppb) average levels of SO$_2$. The LCS SO$_2$ hourly exposure distribution (Fig. 3C) shows an extended high-concentration tail, with 2% of all hourly exposures exceeding 75 ppb (the US EPA 1-h standard, though Hawai‘i uses the pre-2010 threshold of 140 ppb). In comparison, cumulative hourly SO$_2$ exposures from the regulatory network (composed of five stations covering nearly 30,000 people) are 25% of the LCS network’s cumulative total (SI Appendix, Fig. S9). However, the population-weighted average SO$_2$ concentrations measured by the two networks are not significantly different (9.8 ppb and 9.2 ppb) because of highest SO$_2$ concentrations occurring primarily in sparsely populated areas (SI Appendix, Fig. S8).

Fig. 2. Average concentrations of SO$_2$ (A) and PM$_{2.5}$ (B), as measured by the LCS network (colored circles) and the regulatory network stations (gray circles). Data are from a 15-d period from July 15 to August 1, 2018; only the LCS nodes that were in near-continuous operation during this time are shown. For SO$_2$, 17 sensors are shown, accounting for 70,414 people within 5 km. For PM$_{2.5}$, 20 are shown, accounting for 86,856 people within 5 km. In total, there are 16 stations with both SO$_2$ and PM$_{2.5}$ measurements, accounting for 73,013 people within 5 km (Fig. 3). The full time series for all sensors are shown SI Appendix, Figs. S3 and S5.

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regions exposed to high levels of PM$_{2.5}$, cumulative hourly PM$_{2.5}$ exposures from the regulatory network during the study period are 28% of the LCS network (SI Appendix, Fig. S9). Additionally, the population-weighted average PM$_{2.5}$ concentration from the LCS network is substantially higher than that of the regulatory network (12.9 μg m$^{-3}$ to 8.2 μg m$^{-3}$; SI Appendix, Fig. S8), demonstrating the importance of dense monitoring networks to complement existing measurements for accurate population-wide exposure distributions.

The highly variable concentrations (Fig. 2) and exposures (Fig. 3) throughout the island arise not only from differences in location relative to the fissures and the plume but also from the dynamic chemical evolution of the volcanic pollutants (the oxidation of SO$_2$ to form PM). Such secondary transformations of pollutants represent major challenges in AQ monitoring, which spatially dense LCS networks are well-suited to address. The high spatial resolution of our LCS network, with stations placed at different distances (i.e., plume transport times) downwind of the fissure, enables this chemical transformation to be mapped.

Fig. 3. Population exposure to volcanic pollutants, measured by the LCS network over the 15-d study period. (A and B) Mean pollutant distribution as a function of cumulative near-node population (residents living within 5 km of each node: 73,013 total). Bar width is proportional to nearby population, and bar height is the average pollutant concentration measured by each node. Sensor nodes are differentiated by color, as shown on the inset map. Stations are arranged from lowest to highest average concentration. (C and D) Population distribution as a function of hourly exposure frequency to SO$_2$ and PM$_{2.5}$. Here, the distribution of hourly concentrations experienced by each sensor node is weighted by population within 5 km of the node and arranged by average concentration. Estimation of near-node population is given in the SI Appendix. Hourly population-weighted time series data to create (C and D) is shown in SI Appendix, Fig. S9. An equivalent figure using regulatory network data are shown for comparison in SI Appendix, Fig. S10.
aiding in emergency response, measuring populations’ exposure to pollution, and characterizing regional atmospheric chemistry. Key features of this network were the individual nodes’ low cost, small physical footprint, stand-alone power, and real-time communications, all of which allowed for rapid, flexible deployment. Two nodes were even lost in the lava flow, highlighting the resilience of networks composed of multiple low-cost nodes. This environment—with its small number of pollutants, point sources emitting into a clean environment, and relatively simple chemistry and meteorology—is in many ways an ideal scenario for monitoring AQ using LCS. Nonetheless, this general approach can be extended to other, more complex environments as well. The present work highlights the need for multipollutant LCS nodes that are already calibrated and readily deployable; the characterization of other extreme AQ events, such as wildfires and urban smog, requires that these nodes measure a number of additional pollutants (not just PM and SO2 but also O3, NOx, and CO2) as well as species that can provide insight into pollutant sources and secondary chemical processes (such as NOx, VOCs, and CO2). Furthermore, knowledge of background pollutants levels and atmospheric dynamics is necessary to isolate contributions from the event and constrain reaction kinetics. These are included in the present analysis and will be even more important in regions with higher baseline concentrations or more complex meteorology. This underscores the need for improved characterization of regional background pollutant levels (from prior regulatory, satellite, or LCS measurements) and local meteorological conditions (from wind measurements and dispersion models) in such cases. Because of the chemical complexity of wildfire and urban smog pollution, LCS measurements will also benefit from future technological improvements, such as in-line dryers to obviate the need for uncertain relative humidity corrections, and low-cost techniques for measuring PM and VOC composition to provide insight into pollutant sources, chemistry, and impacts. LCS networks thus offer the potential for characterizing pollutant exposure and chemistry under a wide range of conditions and represent an important high-resolution component of multiplatform systems to monitor and characterize extreme AQ events.

Materials and Methods

LCSs and Network Design. Custom multipollutant AQ sensor (MPAQS) sensor nodes were built to measure sulfur dioxide gas (Alphasense SO2-B4 electrochemical sensor) and PM (Alphasense OPC-N2) concentrations, as well as auxiliary measurements of air temperature (T) and relative humidity (RH). Additionally, six SO2-only nodes (24) and five PM-only nodes (Plantower PM5003 nephelometer) were used to supplement the MPAQS network. Communications and data transmission are via 3G cellular microcontroller (Particle Electron). Sensor units are powered by rechargeable batteries and solar panels (Voltaic Systems). All sensors sample at 1 Hz and data are recorded to local storage, with 1-min averages transmitted to a custom cloud database. Total materials cost for each MPAQS node is approximately 1,000 US dollars.

The SO2 and PM sensors were calibrated via colocation outdoors with regulatory-grade monitors run by the Hawai‘i Department of Health (33). SO2 calibration algorithms are based on sensor-specific nonlinear relations to air temperature and linear sensitivity to ambient SO2 concentrations (SI Appendix). PM mass measurements are statistically corrected for ambient RH due to hygroscopic uptake by sulfate aerosols (SI Appendix). This type of field calibration approach is advantageous because instruments are exposed to dynamic and realistic ambient environmental conditions, which can be challenging to achieve under controlled laboratory settings (e.g., ref. 24). Sensor nodes were placed primarily at schools, public health clinics, and community centers in residential areas, and each individual setting and placement was unique according to specific site characteristics and constraints (SI Appendix). Nodes were placed 3 to 15 m above ground level on rooftops (on tripods, electrical masts, etc.) or in open areas on facility grounds (on masts, tree trunks, and utility poles). Although sensors were deployed to more than 30 locations around the Island (SI Appendix), several nodes experienced intermittent communications interruptions, power issues, or sensor component failures (in particular, several SO2 electrochemical

Fig. 4. Chemical evolution of the vog plume (SO2/PM2.5 ratio), as measured by LCS nodes (circles) and regulatory stations (triangles) downwind of the volcano during the July 15 to August 1 study period. This quantifies the rate of the chemical transformation of SO2 (gas) to sulfuric acid (PM), yielding an estimated timescale of 7.6 x 10^-6 s^-1 (lifetime of 36 h). Measurement uncertainties (vertical error bars) were determined separately for SO2 and PM2.5 sensors during instrument calibration against reference instruments (SI Appendix). Plume age uncertainties (horizontal error bars) are the interquartile range of model parcel travel times between the LERZ and measurement location during each hour of the study period.

(see Materials and Methods), this value is consistent with known atmospheric SO2 oxidation kinetics (28) and suggests some role of not only gas phase oxidation but also conversion within cloud droplets or particles as well. Moreover, these reaction kinetics are comparable to those of other volcanoes worldwide (29), within bounds of previous measurements at Kilauea (11, 17, 30–32), and in broad agreement with a recent estimate from the same eruption (17). Previous studies have used a range of measurement approaches—single-point, ground-based SO2 photometer observations (31), time-dependent downwind SO2 and sulfate measurements (30), satellite SO2 measurements (32), and SO2 and PM measurements from AQ stations (17)—which varied substantially in temporal and spatial coverage. The range of lifetimes of SO2 from Kilauea reported in the literature, τ_ox = ~9 h (31) to ~22 d (30), likely results from complex plume SO2 dynamics, the measurement approach used, and differences in atmospheric conditions (e.g., solar radiation, temperature, and moisture) and plume composition (e.g., aerosols, water, and oxidants). An advantage of using LCS measurements in this case is that in situ measurements over multiple days from a relatively large number of spatially distributed stations are used to aggregate highly variable plume meteor-chemical conditions. This study validates the use of distributed LCS networks for monitoring not only air pollutant concentrations but also plume chemical evolution.

In all, the LERZ eruption lasted for ~13 wk before ending abruptly in early August 2018. After the sudden cessation of volcanic activity, AQ conditions improved appreciably across the island as SO2 levels fell immediately and PM2.5 returned to background levels after 7 to 10 d (Fig. 1 C and D). While unexpected, this extreme AQ event provided the opportunity to rapidly implement an LCS network, demonstrating the strengths of LCS for mapping pollution exposure and chemistry during an extreme air quality event (the 2018 Kilauea eruption) using a low-cost sensor network.
sensors failed, likely because of high ambient humidity). In total, there were 16 locations during the 2-wk study period with sufficient data quality and consistency for both SO2 and PM2.5 (Figs. 2 and 3; 16 locations with PM2.5 and SO2 at 1 location with SO2 only, and 4 locations with PM2.5 only).

Population Exposure Data and Analysis. Spatial population data are 2015 Census Designated Place (CDP) level populations from the Hawai'i State Data Center. The CDP polygons are rasterized assuming uniform density within each CDP (SI Appendix). The data are based on night-time residential population, and there are likely uncertainties based on nonresident populations (e.g., tourists and temporary workers), diurnal commuter patterns, and spatial settlement patterns (e.g., variations in housing density and informal encampments).

In Fig. 3, the 5-km buffer is intended as a compromise between spatial variability and representativeness of point measurements and population coverage. To avoid population double counting when stations are within 5 km, one half of the total number of residents living in the overlapping buffer zone is assigned to each station.

There are two methods used to calculate the 5-km population-weighted exposures shown in Fig. 3. First, total population exposure (x) for each node (i) (Fig. 3 A and B) is calculated by multiplying the number of residents within 5 km of the sensor node (pi) by the mean observed concentrations ([c] of SO2 (person ppb) and PM2.5 (person μg m⁻³); see SI Appendix). Second, time-integrated exposure distributions (Fig. 3 C and D and SI Appendix, Fig. S9) for each node (xi) are calculated as the summed product of population and smoothed (three-bin moving window) binned frequency counts (j) of hourly SO2 (person ppb hours) and PM2.5 concentrations (person μg m⁻³ hours):

\[ x_{i} = \sum_{j=1}^{n} p_{i} [c]_{j} \]

SO2 to PM Conversion Rate. The mean SO2 reaction rate is calculated based on observed SO2 and PM2.5 at 13 stations downwind of the eruption. The SO2 and PM2.5 concentrations are converted to mass concentrations of sulfur (S in units of μg m⁻³) and the fraction of total gas phase (f[S]) is

\[ f_{S} = \frac{S_{SO2}}{S_{SO2} + S_{PM2.5}} \]

This procedure assumes all remaining PM2.5 is secondary sulfuric acid aerosols, originating from volcanic SO2 emissions. (The mass concentration of liquid water had been subtracted already as part of the PM RH-correction; SI Appendix.) This approach neglects background levels of nonvolcanic aerosols (~5 μg m⁻³ based on the network average value of SO2 measured from September 1 to 30, 2018 after the eruption had ceased). This fractional conversion is advantageous because reaction rates calculated using SO2 measurements alone can be overestimated due to nonoxidative SO2 losses from deposition or dilution (30).

An exponential decay function (λ) is then fit to the downwind measurement points:

\[ f_{S} = f_{0}e^{-\lambda t} \]

where t is the mean plume travel time between the LERZ and measurement locations from July 15 to August 1, 2018, calculated using a particle dispersion meteorological model (SI Appendix). Here, kSO2 represents the first-order decay constant of SO2 and so the mean SO2 lifetime is equal to kSO2⁻¹. To incorporate measurement and plume age model uncertainties into confidence and prediction intervals (95%) (Fig. 4), the decay function is fit to an array of random points uniformly sampled within the uncertainty bounds at each measurement point (n = 10 points at each location; total n = 130). Measurement uncertainties were determined for SO2 and PM2.5 sensors during instrument calibration against reference instruments (SO2 mean absolute error [MAE] is 7.3 ppb and PM2.5 MAE is 4.5 μg m⁻³) and modeled plume age uncertainties are the interquartile range of calculated parcel travel times between the LERZ and measurement location during each hour of the study period (SI Appendix).

The plume SO2 conversion rate (kSO2) fit to all downwind stations (n = 13) is 7.6 × 10⁻⁶ s⁻¹ (t50 = 36.3 h), while the kSO2 fit to only the LCS network stations (n = 10) is 7.2 × 10⁻⁶ s⁻¹ (t50 = 38.3 h), a statistically insignificant difference demonstrating the ability of the low-cost network to observe reaction kinetics. Additionally, there is not a dramatic difference when kSO2 is fitted to measurements that have had approximate background PM2.5 levels subtracted (kSO2 = 6.0 × 10⁻⁶ s⁻¹; t50 = 46.4 h). However, kSO2 fit using plume ages estimated from an observed mean wind speed at a single location and downwind linear distance is 1.9 × 10⁻⁶ s⁻¹ (t50 = 15 h), substantially faster than the rate using plume ages calculated from the particle dispersion model. This demonstrates the importance of local meteorological variations, especially in areas of complex terrain, to estimate reaction kinetics.

Data Availability. All study data are included in the article and/or SI Appendix.

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2. R. V. Martin et al., No one knows which city has the highest concentration of fine particulate matter. Atmos. Environ. X 3, 100040 (2019).


